

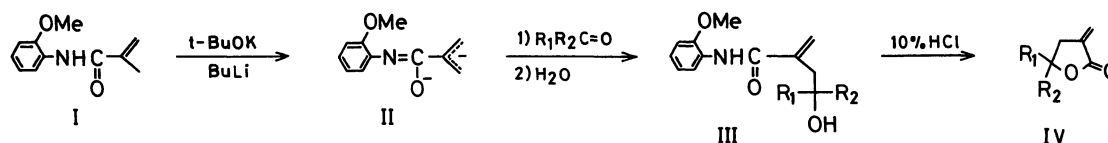
DIANION OF *N*-(*o*-METHOXYPHENYL)-2-METHYLPROPENAMIDE:  
 A NEW REAGENT FOR THE CONVENIENT SYNTHESIS OF  $\alpha$ -METHYLENE- $\gamma$ -  
 BUTYROLACTONES FROM CARBONYL COMPOUNDS

Kazuhiko TANAKA\*, Yoshihito NOZAKI, Norikazu TAMURA,  
 Rikuhei TANIKAGA, and Aritsune KAJI  
 Department of Chemistry, Faculty of Science,  
 Kyoto University, Sakyo-ku, Kyoto 606

Dianion of *N*-(*o*-methoxyphenyl)-2-methylpropenamide was successfully generated on treatment of the amide with *t*-BuOK-BuLi at  $-78^{\circ}\text{C}$  in THF and utilized as the key reagent for the synthesis of  $\gamma$ -substituted  $\alpha$ -methylene- $\gamma$ -butyrolactones.

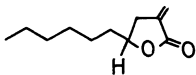
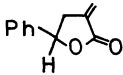
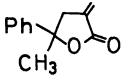
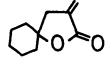
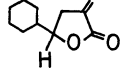
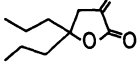
The conversion of carbonyl compounds into  $\alpha$ -methylene- $\gamma$ -butyrolactones is an important synthetic process which has received considerable recent attention.<sup>1)</sup> Although several procedures have been developed for such transformation, there is no general method for the synthesis of  $\alpha$ -methylene- $\gamma$ -butyrolactones using carbanions derived from  $\alpha,\beta$ -unsaturated amides.<sup>2)</sup> The generation of the stable carbanions from  $\alpha,\beta$ -unsaturated amides is difficult owing to the tendency of these carbanions to undergo facile Michael addition to the starting amides.<sup>3)</sup>

We report herein the successful generation of the stable carbanion of *N*-(*o*-methoxyphenyl)-2-methylpropenamide (I) and its application to the synthesis of  $\gamma$ -substituted  $\alpha$ -methylene- $\gamma$ -butyrolactones.<sup>4,5)</sup>



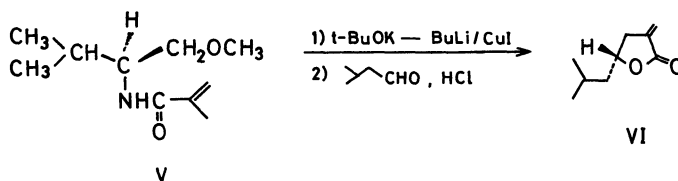
A typical procedure for the conversion of acetophenone into  $\gamma$ -methyl- $\alpha$ -methylene- $\gamma$ -phenyl- $\gamma$ -butyrolactone (IVc) is as follows. To a suspension of *t*-BuOK (44 mmol) in dry THF (60 ml) at  $-78^{\circ}\text{C}$  under  $\text{N}_2$  was added a solution of I (20 mmol) in 5 ml of THF. After 10 min, butyllithium (44 mmol) was added and the resulting orange suspension was stirred for 30 min. Acetophenone (22 mmol) in 5 ml of THF was then added. Product isolation and purification by chromatography gave a 59% yield of IIIc as a viscous oil;  $^1\text{H-NMR}(\text{CCl}_4)$   $\delta$  8.30(m, 2H), 7.10(m, 8H), 5.55(s, 1H), 4.98(s, 1H), 4.95(m, 1H), 3.80(s, 3H), 2.76(m, 2H), 1.53(s, 3H). Hydrolysis of IIIc in 10% hydrochloric acid afforded an 89% yield of IVc;  $^1\text{H-NMR}(\text{CDCl}_3)$   $\delta$  7.30(m, 5H), 6.16(t, 1H), 5.55(t, 1H), 3.13(t, 2H), 1.70(s, 3H); IR(neat) 1752(COO), 1660(C=C)  $\text{cm}^{-1}$ . Anal. Found: C, 76.52; H, 6.59%. Calcd for  $\text{C}_{12}\text{H}_{12}\text{O}_2$ : C, 76.57; H, 6.44%. Other examples of this new procedure are listed in

Table 1. Synthesis of  $\alpha$ -methylene- $\gamma$ -butyrolactones

Carbonyl compound	III, Yield(%) <sup>a)</sup>	IV, Yield(%) <sup>a,b)</sup>	Mp. °C
Heptanal	57 (IIIa)	 (IVa) 88	oil <sup>c)</sup>
Benzaldehyde	46 (IIIb)	 (IVb) 80	54-55 <sup>d)</sup>
Acetophenone	59 (IIIc)	 (IVc) 89	oil <sup>c)</sup>
Cyclohexanone	55 (III d)	 (IVd) 87	oil <sup>c)</sup>
Cyclohexanecarbaldehyde	45 (III e)	 (IVe) 88	55-56
4-Heptanone	56 (III f)	 (IVf) 88	oil <sup>c)</sup>

- a) Isolated yields. b) All products gave satisfactory analytical results and spectral data. c) Purified by column chromatography since distillation led to decomposition. d) Lit.<sup>2)</sup> 49-54°C.

Table 1. This strategy can also be applied to the asymmetric synthesis of  $\alpha$ -methylene- $\gamma$ -butyrolactone. The chiral amide (V) was conveniently prepared from L-valine. 3-Methylbutanal was treated with the dianion prepared from V in the presence of 1.1 equiv of CuI and the resulting adduct was then heated under reflux in 10% hydrochloric acid. The desired lactone VI was thus obtained in 16% overall yield (12% optical purity).<sup>6)</sup>



Acknowledgment: Support for this research by the Ministry of Education is gratefully acknowledged.

## References and Notes

- 1) P. A. Grieco, *Synthesis*, **1975**, 67; R. B. Gammill, C. A. Wilson, and T. A. Bryson, *Synth. Commun.*, **5**, 245 (1975).
- 2) E. Öhler, K. Reininger, and U. Schmidt, *Angew. Chem.*, **82**, 480 (1970); L. S. Hegedus, S. D. Wagner, E. L. Waterman, and K. Shiirala-Hansen, *J. Org. Chem.*, **40**, 593 (1975); R. M. Carlson and A. R. Oyler, *J. Org. Chem.*, **41**, 4065 (1976); R. M. Carlson, *Tetrahedron Lett.*, **1978**, 111.
- 3) W. Bannwarth, R. Eidenschink, and T. Kauffmann, *Angew. Chem. Int. Ed. Engl.*, **13**, 468 (1974); J. E. Baldwin and W. A. Dupont, *Tetrahedron Lett.*, **1980**, 1881.
- 4) A report describing the  $\beta'$ -lithiation of  $\alpha,\beta$ -unsaturated amides has recently appeared. P. Beak and D. J. Kempf, *J. Am. Chem. Soc.*, **102**, 4550 (1980); see also J. A. Oakleaf, M. T. Thomas, A. Wu, and V. Snieckus, *Tetrahedron Lett.*, **1978**, 1645.
- 5) Reported in April 1980 at the Meeting of the Japan Chemical Society, Osaka. Abstracts II, p 1023.
- 6) Based on  $[\alpha]_D +67.0^\circ$  (c 1.44, EtOH); see K. Mori, *Tetrahedron*, **32**, 1101 (1976).

(Received September 8, 1980)